Bharati Vidyapeeth's Dr. Patangrao Kadam Mahavidyalaya, Sangli Internal Quality Assurance Cell Department of Chemistry

PROGRAMME OUTCOMES

Name of Programme: M. Sc. Analytical Chemistry

PO1: The M.Sc. analytical chemistry program at Shivaji University, Kolhapur provides the key knowledge base and laboratory resources to prepare students for careers as professionals in the field of chemistry and particularly in analytical chemistry enabling them to interface not only with various branches of chemistry (organic, inorganic, physical, biological, industrial, environmental, pharmaceuticals etc) but also with the related fields, and for professional courses and areas of research including medical, forensic, food, agriculture, dental, law, intellectual property, business programs etc.

PO2: Students will be able to solve various problems by identifying the essential parts of a problem, formulate strategy for solving the problem, applying appropriate techniques to arrive at a solution, test the precision and accuracy of the solution and interpret the results.

PO3: Students will be able to acquire domain specific knowledge and technical skills needed for employment in industries, teaching fields and pursue research. Students will be skilled in problem solving, critical thinking and analytical reasoning

PO4: Students will be able to apply the fundamental knowledge to address the crosscutting issues such as sustainable development

PO5: Students will get perfect insight into qualitative and quantitative analytical chemistry and research ethics for production of quality research.

PO6: Students will be able to communicate effectively i.e. being able to articulate, comprehend and write effective reports, make effective presentations and documentation and capable of expressing the subject through technical writing as well as through oral presentation.

PROGRAMME SPECIFIC OUTCOMES

Name of Programme: M. Sc. Analytical Chemistry

PSO1: Students will be able to prepare and qualify subject specific competitive exams like NET, SET and GATE and also other general public administration exams like M.P.S.C. and U.P.S.C. etc. exams.

PSO2: Student will be able to utilize the knowledge and analytical skills in QA-QC and R&D departments in almost all the industries enabling them to secure jobs where analytical chemistry is the core requirement to ensure and ascertain the quality of the product.

PSO3: Students will have opportunity for higher education leading to Ph.D. program.

PSO4: Students will be able to explore contemporary research in chemistry and allied fields of science and technology, collaborate in team projects, communicate the results of scientific work in oral, written and electronic formats to both scientists and the public at large.

PSO5: Students can start their own laboratories/startups/ chemical industry/ business (entrepreneurship).

PSO6: Students will be able to interpret data from the state of art Analytical instruments for ascertaining the product/material.

Course Outcomes

Course	Outcomes
OCH 101- (Organic Chemistry-I)	After completion of these courses, students should be able to,CO-1: To learn and understand types of reactions, structure, stability and reactivity of carbenes, arynes ,nitrenes, and SN1, SN2, SNi reactions.CO-2: To learn three, four and five membered system.
ICH 102- (Inorganic Chemistry-I)	 CO-1: To understand CFT for Td, Oh, Sq. planar, TBP filled, CFSE and its applications. CO-2: To learn the classification, nomenclature, synthesis, bonding and properties of organometalic compounds. CO-3: To learn preparation, structure and properties of metal carbonyls. CO-4: To understand symmetry, point groups an Milliken symbolism rules.
E-ACH103- (Analytical Chemistry I) Elective paper	 CO-1: To understand TGA, DTA, DTG and DSC and its applications. CO-2: To learn AAS in detail and its application, FES, ICP and its application. CO-3: To understand Beer-Lambert's law, structural problems and its application. CO-4: To understand IR spectroscopy in detail, problems and its applications.
CH104- (Research Methodology)	 CO-1: To understand Research methodology and implementation of research. CO-2: To learn online searching, impact factor and paper writing for international journals. CO-3: To understand Errors, accuracy, precision. CO-4: To learn how to operate PC and How to learn standerd programs.

	M.Sc.I, Sem-I (Chemistry Practical)		
Course	Outcomes		
Laboratory practical	After completion of these courses, students should		
	be able to,		
	CO-1: To learn ore, alloy analysis.		
	CO-2: To learn preparation of coordination complexes		
	CO-3: To learn instrumentation techniques.		
	CO-4: To learn single stage preparation of important		
	organic products.		
	CO-5: To learn rate of reaction, kinetics of reaction.		
	CO-6: To learn estimation and preparation of organic		
	compounds.		
M. Sc. I (NEP-2020) Sen	nester II		
Course	Outcomes		
	After completion of these courses, students should		
PCH 201-	be able to,		
(Physical Chemistry-II)	CO-1: To understand wave functions, spectroscopic term		
	symbols and numericals.		
	CO-2: To understand partition function, thermodynamic		
	properties and numericals.		
	CO-3: To know the theory of strong electrolyte, Debye		
	Hukkel theory and numericals. CO-4: To learn kinetics of reaction and steady state		
	approximation.		
ACH 202-	CO-1: To learn basics of analysis, statistics in chemical		
(Analytical	analysis and MS office in chemistry applications.		
Chemistry-II)	CO-2: To understand volumetric and gravimetric		
	analysis.		
	CO-3: To understand Gas, HPLC and Ion exchange		
	chromatography.		
	CO-4 To learn Voltametry, polarography, amperometry		
	and electrogravimetry analysis.		
	CO 1 The second stand NIMP and Instance station of PT		
E-ACH203-	CO-1: To understand NMR and Instrumentation of FT-		
(Analytical	NMR and its applications.		
Chemistry)	CO-2: To learn types of ionization, analyzers and		
Elective paper	application's. CO-3: To learn Rigid and non-rigid rotors and		
	numericals.		
	CO-4: To understand Raman Spectra, Vibrational		
	Raman spectra and numericals.		

Course Outcomes M.Sc.I, Sem-II (Chemistry Practical)		
Course	Outcomes	
Laboratory practical	After completion of these courses, students should be	
	able to,	
	Co-1: To learn ore, alloy analysis	
	CO-2: To learn preparation of coordination complexes .	
	CO-3: To learn instrumentation techniques.	
	CO-4: To learn single stage preparation of important	
	organic products.	
	CO-5: To learn rate of reaction, kinetics of reaction.	
	CO-6: To learn estimation and preparation of organic	
	compounds.	

M. Sc. II (NEP-2020) Semester III

ACH-	(Advanced Analytical	CO1: Develop knowledge of fundamental,
3.1	Techniques)	instrumentation and working of state of art
		instrumental analytical techniques, effective use
		and choice of technique, written and/or oral
		communication of the concepts of analytical
		chemistry which will be useful as analytical chemist
		and R&D. CO2: Acquire knowledge of mass
		spectrometry, type of MS, ionization types and
		specific practical applications of MS.
		CO3: Acquire knowledge of basics of nanochemistry,
		nanomaterials and nanotechnology and application
		orientated synthesis and characterization of
		nanomaterials.
		CO4: This course gives wide understanding about
		the instrumental analytical techniques (SEM, TEM,
		EDS, STM, AFM, Raman, XFS, ESR, XPS, AES, SIMS
		etc.)employed for qualitative and quantitative
		analysis for contemporary research.
ACH-	(Organic Analytical	CO1: Students will gain knowledge of the
3.2	Chemistry)	instruments used at the interface of Analytical- Organic chemistry useful for R&D and structural
		elucidation using UV-Visible, IR, 1H & 13C NMR,
		Mass spectrometry data and interpretation of the
		same.
		CO 2: Students will acquire knowledge about the
		drug, their classification, sources of impurities
		(chemical, atmospheric and microbial
		contamination) in pharmaceutical raw materials
		and analysis of the same.
		CO 3: Students will gain knowledge about the

		
		conventional and advanced analytical approaches
		for analysis of drug, vitamin, body fluids and clinical
		samples. CO 4: Students will have an idea of commonly used
		pesticides and their analysis and also about forensic
		science and forensic sample analysis.
ACH-	(Electroanalytical	CO1: Fundamental knowledge of electrochemistry,
3.3:	Techniques in Chemical	electrodes, types of electrodes, its construction will
0.01	Analysis)	lay foundation for the course.
	indiy 515 j	CO2: Students will gain knowledge and skill in
		electroanalytical techniques like cyclic
		voltammetry and its types, polarography,
		coulometry and dynamic light scattering
		technique for qualitative and quantitative
		analysis.
		CO3: Students will be familiar with the advanced
		electrodes used for chemical analysis, liquid-liquid
		membrane electrodes, enzymes and gas
		electrodes.
		CO4: Students will learn about electrophoretic
		techniques, advances in electrophoresis
		techniques and its analytical applications.
ACH-3.		CO1: Students will acquire knowledge about
4)	(A)	sampling, criteria of good sampling, handling,
,	(Environmental	preservation and storage of the samples,
	Chemical Analysis and	pretreatment and post treatment of samples.
	Control)	CO2: Students will acquire knowledge of conditions
		and strategies required during sampling and
		electrochemical and spectral methods for analysis of environmental samples.
		CO3: Students will learn about the air and water
		pollution, sources of pollution, typical parameters
		and properties (physical, chemical and biological)
		to be measured in air and water pollution with
		relevance to specific case studies.
		CO4: Students will be acquainted with organic
		pollutants and their analysis with special reference
АСН-	(B) (Docont Advances in	to pesticide analysis.
асн- 3.4)	(B) (Recent Advances in Analytical Chomistry)	CO1: Students will be acquainted with ultra-purity
3. 4 J	Analytical Chemistry)	and ultratrace analysis required in electronic and semiconductor processing.
		CO2: Students will learn Radio-Analytical
		techniques for analysis.
		CO3: Student will be well versed with C13, P15 and
		017 NMR Spectroscopy applications.
		CO4: Student will learn about ESR spectrometry and
		its applications quantitative analysis.
ACH-	(B) (Recent Advances in	CO1: Students will be acquainted with ultra purity
3.4	Analytical Chemistry)	and ultra trace analysis required in electronic and
011		semiconductor processing.
L		semiconductor processing.

		CO2. Studente will leave Dedie Arelatical
		CO2: Students will learn Radio-Analytical
		techniques for analysis.
		CO3: Student will be well versed with C13, P15 and
		017 NMR Spectroscopy applications.
		CO4: Student will learn about ESR spectrometry and
		its applications quantitative analysis.
ACHP -	Practical -V	CO1: In-depth training on laboratory solution
V		preparations on all concentration scales
		CO2: Training on laboratory safety and lab ethics in
		scientific work
		CO3: Training on planning, design and execution of
		experiments
		CO4: Training on uncertainty estimations for
		experimentally measured and derived properties of
		solutions
ACHP -	Practical-VI	CO1: Training on scientific literature search,
VI		defining the objective of the work, research skills,
		data representation in tabular and graphical form
		etc.
		CO2: Training on experimental verification of
		fundamental theories, comparison of data with
		literature and scientific discussion on any deviation
		of data from expected theoretical values or reported literature.
		CO3: Developing analytical skills
		CO4: Training on qualitative and quantitative
		analysis of analyte
	Part-II semester-IV	
ACH4.1	(Modern Separation	CO1: Students will learn about modern separation
	Method in Analysis)	and chromatographic used for analysis of different
		type of samples.
		CO2: The student will understand instrumentation
		and mechanism of various separation techniques.
		CO3: Student will acquire knowledge regarding
		various choice of instrument and detectors to be
		used for analysis depending on the sample and
		matrix.
		CO4: Student will learn fundamentals of extractive
		chromatography, types of extraction techniques,
		advances in extraction methods and their
		hyphenations with chromatography leading to
		addressing challenging problems in analytical
		chemistry.
ACH-	(Organic Industrial	CO1: Acquire knowledge of handling and
4.2	Analysis)	investigating the characteristics of the oils, fats,
		detergents and soap samples and analysis of the
		same providing opportunity in cosmetic,
		pharmaceuticals, dyes and polymers industries.
1	1	$p_{11}a_{11}a_{12}a_{13}a_{1$

		CO2: Student will gain knowledge and importance of food quality, probe for food adulteration and adulterants, food preservative, food flavors and analysis of their components. CO3: Students will also gain knowledge about the animal food stuff and the additives added in the animal food stuff as antibiotics, dietary supplements and growth promoting drugs, preservatives etc. and analysis of the same. CO4: Student will learn about the analysis of cosmetics, face powder, hair dyes and hair care products, types of cosmetics, precautionary measures and composition of the cosmetics and specific roles of the ingredients. Will acquire knowledge about the paints, pigments and petroleum products, composition and analysis of the same using conventional and instrumental techniques.
ACH- 4.3	(Advanced Methods in Chemical Analysis)	CO1: Students will be skilled in the techniques like fluorescence, phosphorescence, types of quenching, FRET and applications of the same in Analytical Chemistry and for addressing research problems. CO2: Students will gain knowledge of the kinetic methods of analysis supporting the analysis and data procured in research. CO3: The students will acquire the knowledge of advanced method of chemical analysis XPS, XRF, fluorescence and phosphorescence spectroscopy which will be beneficial in research. CO4: Students will acquire knowledge of identifying types of plastic and will also be able to and
ACH- 4.4 (A)	(Industrial Analytical Chemistry)	determination of metallic impurities in plastics CO1: The students will acquire knowledge of analysis of metals, alloys, minerals and ores commonly used in the industry. CO2: The students will be acquainted with the analysis of real samples like cement, plaster of Paris, different commercial ores, soil composition, soil fertility, fertilizers etc using conventional and instrumental methods of analysis. CO3: Students will also gain the knowledge of analysis of commercial materials, explosives, polymers, resins, rubber, luminescent paints, lubricants and adhesives. CO4: These would offer opportunity to the students to get employment in industries for quality assurance and quality control (QA-QC) of the product.

ACH-	(Quality Assurance and	CO1: Students will acquire knowledge of QA-QC
4.4 (B)	Accreditation)	which in essential for analytical chemist, This covers
(-)		a variety of chemical fields and this knowledge
		would help students working on various materials,
		understanding the basics of samples, sampling,
		sample storage, and pre-post treatment of samples.
		CO2: Students will acquire knowledge of good
		laboratory practices, professional ethics, and
		instrumental analytical chemistry, awareness of health hazards, remedial measures, analytical
		method development and validation.
		CO3: The students would be aware of the
		importance of documentation for raw materials and
		finished products, their monitoring, maintenance
		and management.
		World-wide agencies involved in regulating the
		analytical protocols and establishing standards. CO4: Students will gain knowledge about the quality
		assurance and accreditation, evolution and
		significance of quality management, available
		accreditation agencies and advantages of
		accreditation.
ACHP –	Practical-VIII	CO1: The students will acquire hands on training for
VIII		conducting the representative experiments for the
		analysis of wide variety of samples of inorganic, organic and physical approaches by qualitative and
		quantitative analysis. Demonstrate professional and
		ethical attitude to serve the society
		CO2: Students will have knowledge of safety signs
		on container of chemicals, safety in handling of
		chemicals, MSDS sheets, learn sample preparation
		and characterization for confirming the purity. CO3: Students would acquire knowledge about the
		separation and estimation of amount of metal, metal
		ions, organic compounds etc. in given samples.
		CO4: Based on the experience of project work,
		students will have ability to start their R & D
		laboratory.